## AMENDED SPECIFICATION

Reprinted as amended in accordance with the Decision of the Superintending Examiner acting for the Comptroller-General dated the eleventh day of October, 1957, under Section 33, of the Patents Act, 1949.

> SPECIFICATION PATENT



Date of Application and filing Complete Specification: Oct. 13, 1949.

No. 26316/49.

Application mode in United States of America on Nov. 23, 1948.

Complete Specification Published: Jan. 7, 1953.

Index at acceptance:—Class 2(iii), B4a(2:4), B4(d:e), Cla(1:10), Clb(1:2), Cle4k(3:4:6:8), Cle5k(4:6:8), Cle6k(4:6:8), Clf2a(2:3), Clf2c(4:5:6), Clf2d(1:2:3), C2a(3:5:14), C2b3 (a4:b:f), C2b3g(1:7), C2b(4:9), C(2r17:3a:12), C3a13c(a4:b3:f3), C3a13c(lc:2c:3c:6c:9), C3a13c10(f:h), C3a14a(3d:fa2) 5:8d).

## COMPLETE SPECIFICATION

## Process for the Manufacture of Anti-Histaminic Compounds

We, Schering Corporation, having a place of business at 2, Broad Street, Bloom-field, County of Essex, State of New Jersey, United States of America, a corporation organized under the laws of the State of New Jersey, United States of America, (Assignee of NATHAN SPERBER, residing in Bronx, County of Bronx, State of New York, United States of America, and DOMENICK PAPA, residing in Brooklyn, County of Kings, State of New York, United States of America, both Citizens of the United States of America), do hereby declare the nature of this invention and in what manner the same is to be per-15 formed, to be particularly described and ascertained in and by the following statement: -

This invention relates to new substances of interesting and important physiological 20 properties and a process for their manufacture. More specifically, the invention relates to the preparation of compounds having pronounced antihistaminic activity.

It is recognized that the liberation of hist-25 amine into the tissues, which can be brought about by a multitude of agents or processes, is primarily responsible for many of the allergicmanifestations in man. It has been found that certain substances of closely reliated chemical configurations are effective in alleviating the symptoms of many allergic reactions. The specificity of these chemical substances for the control of allergic reactions is well demonstrated by the researches carried on within the last ten years. However, although the substances prescribed at the present time represent a remarkable advance they exhibit many undesirable side effects, or so-called toxic reactions, among which may be mentioned the high incidence of drowsiness, dizziness, nausca, gastro-intestinal irritation and dryness of the mouth.

In specifications Nos. 307,304 and 646,198 (both as open to public inspection under Section 91 of the Patents Acts 1907-1946) general methods are described for the conversion of kerones of the formula:

R'.CO.X.NR'R'

by Grignard reaction into carbinols:

and in Specification No. 646,198 for the subsequent replacement of the hydroxyl group by hydrogen, R<sup>1</sup>, R<sup>2</sup>, R<sup>2</sup> and R<sup>4</sup> being monovalent organic radicals (NR<sup>2</sup>R<sup>3</sup> may be a nitrogen ring residue) and X being a divatent linking 55 group. The products are stated to have good musculotropic antispasmodic activity accompanied by low neurotropic antispasmodic activity. In Specification No. 646,198 as open to public inspection under Section 91 N-(3phenyl - 3-cyclohexylpropyl)-piperidinehydrochloride, obtained in this manner from Npiperidylpropiophenone and cyclohexyl bromide, is said to have 12 times the musculotropic antispasmodic activity of papaverine.
We have now found that certain com-

pounds obtainable by similar general reactions possess to an ourstanding degree anti-histaminic and antianaphylactic activity. Particularly important is the comparative absence of any sectation, dizziness or depression in more than 90% of the cases treated. This advantage is of extreme importance in the clinical application of antihistaminic drugs.

The selected compounds showing this ad-10 vantage have the general formula

wherein Py stands for 2-pyridyl, Ar for phenyl or an alkyl-, alkoxy-, dialkylamino, chloro- or brumophenyl or for 2-thienyl, and 15 R for a dialkylamino-, piperidino-, pyrrolidino-, or morpholino-group.

Throughout this specification the terms alkyl and alkoxy are used to denote groups having less than seven carbon atoms.

The compounds of the invention are produced by a process comprising the step of condensing a ketone Ar.CO.CH., CH.R., with an organometallic 2-pyridyl compound (e.g. 2-pyridyllithium or 2-pyridyl magnesium halide) to give the carbinol

followed by replacement of the hydroxyl group by a hydrogen arom. The resulting bases may be converted into their salts by the usual methods.

Thus from  $\beta$  - dimethylaminopropiophenone (I)

there is obtained 1-(21-pyridyl)-1-Phenyl-3-35 dimethylaminopropanol-1 (II):

The carbinol (II) may be reacted with thionyl chloride to form the chloro-compound (III):

which on reduction with zinc dust and aceric

acid gives in good yield the desired 1-phenyl-1-(2'-pyridyl)-3-dimethylaminopropane (IV)

By a similar series of reactions compounds in which the phenyl group carries alkyl, alkoxyl, dialkylamino, chlorine or bromine substituents may be prepared. For the p-chloro-compound, for example the starting material is the ketone

p-Cl.C.H., CO.CH., CH., N(CH.), obtained by the Mannich reaction from p-chloroacetophenone, dimethylamine and formaldehyde.

By using diethylamine, piperidine, pyrrolidine or morpholine in place of dimethylamine the corresponding diethylamino-, piperidino-, pyrrolidino- or morpholino- ketone may be

prepared.

The compounds of the invention may be used in the form of the free bases or in the form of the salts thereof with inorganic acids such as hydrochloric, hydrobromic, sulphuric and phosphoric acids and organic acids, such as salicylic, tartaric, maleic, succinic, cirric and lacric acids.

Typical examples of salts of the 3-phenyl-3-(2-pyridyl) - N,N - dimethypropylamine of Example I are the following:

1. The mono-hydrochloride is chrained by passing anhydrous hydrogen chloride into an ether solution of the γ-phenyl-γ-(2-pyridyl)-N.N-dimethylpropylamine. The hydrochloride can be recrystallized from absolute alcohol and absolute ether and melts at 117—119° C.

2. The tartrate of the compound of Example I is obtained in the usual manner and melts at 114—115° C.

3. The mono-hydrogen oxalate is prepared in echanol and after recrystallization from acctone melts at 152—152.5° C.

4. The mono-hydrogen succinate is prepared in a manner similar to the mono-hydrogen oxalate in ethyl alcohol solution and after recrystallization from pentancl melts at 99.5— 100° C.

5. The mono-hydrogen maleate is similarly prepared and after recrystallization from pensanol, melts at 106—107° C.

The compounds may be used in a variety of forms such as tablets for oral administration, creams for topical application, and injectible solutions. Preferably the salts of the compounds are used in the creams which may be of the usual formulations. The injectible 95 solutions comprise non-toxic salts.

EXAMPLE I.
1-Phenyl-1-(2'-pyridyl) - 3 - dimethylaminopropane.
The intermediate carbinol, phenyl - (2'- 100)

pyridyl)- $\beta$ -dimethylaminoethylcarbinol (II), is Mannich reaction followed by the 2-pyridylprepared as follows: lithium reaction and the sories of reactions desombed in Example I the corresponding B-Dimethylaminopropiophenone hydrochloride (0.1 mole) is dissolved in 50 cc. of water propylamine is prepared; b.p. 139-141 and cooled in an ice-bath. The free base is  $C_{\rm c}/1.0$  mm. liberated with ice and 10% sodium carbonate EXAMPLE IV. 1-(Phenyl)-1-(21-pyzidyl)-3-diethylaminosolution, and the oil is taken up in other. The either layer is washed with water and dried By substituting & diethylaminopropiophenover anhydrous potassium carbonate. Upon reone hydrochloride for the dimethylamino commoval of the ether, the free base is obvained. pound in Example I there is obtained the A solution of 0.2 moles of 2-pyridyllithium compound of this example; b.p. 156-157 in 250 ml. of ether is prepared and after cool-C./2.0 mm. ing to  $-40^{\circ}$  C., a solution of 18 g. of  $\beta$ -dimethylaminopropiophenone in 50 cc. of ether EXAMPLE V. 1-(Phenyl)-1 - (2'-pyridyl)-3-N - piperidino-15 is added dropwise with stirring over a period of & hour. Upon completion of the reaction, By substituting piperidine hydrochloride for the temperature is allowed to rise to -15' dimethylamine hydrochloride in Example I, C. and the reaction mixture is stirred at this the piperidino compound is obtained as a visremperature for one hour. The contents of the cous yellow liquid boiling at 176—177° C/3.5 mm. flasks are decomposed with ice and hydrochloric acid and then made basic with gaseous Example VI. ammonia. The resulting oil is taken up in ... 1-Phenyl-1-(2'-pyridyl)-3-(N-pyrrolidyl)... ether, the ether evaporated and the residue propene. The carbinol is a viscous, yellow distilled. The  $\beta$ -(1-pyrrolidy1)propiophenone is obsyrup, boiling at 176-180° C./2 mm. tained by the Mannich condensation of accto-The carbinol (II) is converted to the phenone with formaldehyde and pyrrolidine. propylamine as follows: The free base is liberated from the hydro-Phenyl-(2'-pyridyl) - \( \beta \) - dimethylaminochloride and then is reacted with 2-pyridylethyl carbinol (II) (0.1 mole) is dissolved in lithium, followed by further reactions in 250 cc. of dry benzene and thionyl chloride accordance with the procedure of Example L (0.15 mole) added, keeping the temperature The pyrrolidylpropane is obtained as a pale between 0 and 10° C. The reaction is allowed yellow oil boiling at 164-166° C./2-3 mm. to come to room temperature, stirred for an additional & hour, and then made basic with EXAMPLE VII. a dilute solution of sodium hydroxide. The 1-(p-Chlorophenyl)-1-(21-pyridyl)-3-(Nbenzene layer is separated, dried and conpyrrolidyl)propane. centrated in vacuo leaving a viscous, purple This compound is obtained exactly as deoil. The crude phenyl-(21-pyridyl)-#-discribed for the unsubstituted compound of the methylaminoethyl-methylchloride is dissolved above example using p-chloroacetophenone in place of acetophenone. The halogenated comin 200 cc. of glacial acetic acid and zinc dust (0.3 mole) added. The reaction mixture is pound of this example is a yellowish liquid stirred and brated on the steam bath for 6 boiling at 175—177° C./1—2 mm. hours, the zinc sains filtered and the filtrate The following are other typical amines preconcentrated in vacuo. The thick syrup is pared by the methods of the invention: made alkaline with dilute sodium hydroxide 1-(2'-Thienyl)-1-(21'-pyridyl)-3 - dimethyl- 110 and the oil which separates is extracted with aminopropane, b.p. 154° C./2 mm. ether. The other layer is dried, concentrated 1-(p-Methylphenyl)-1-(21-pyridyl) - 3 - diand the residue distilled. methylaminopropane, b.p. 137-140° C./0.5 EXAMPLE II. 50 1(p-Methoxyphenyl)-1 - (2<sup>1</sup> - pyridyl) - 3-di-1-(42 - Dimethylaminophenyi) - 1 - (211- 115 methylaminopropane yridyl)-3-dimethylaminopropane, b.p. 183-This compound is prepared by the pro-185° C./1.5 mm. cedure described in Example I using p-meth-1-(21,31-Dimethoxyphenyl)-1 - (211-pyridyl)oxyacetophenone in a Mannich condensation -dimethylaminopropane, b.p. 195-53 with formaldehyde and dimethylamine hydrochloride to prepare B-dimethylamino-p-meth-C./1--2 mm.120  $1-(p-1sopropylphenyl)-1-(2^2-pyridyl)$ oxypropiophenome. The latter is then carried dimethylaminopropane, b.p. 147-152° C/ through the series of reactions described in 1.0 mm. Example I. The substituted propylamine is a pale vellow, viscous liquid; b.p. 172—175° C/1.5 mm. What we claim is:-1. The step in the production of pyridyl 125 aliphatic amines and their sales which con-EXAMPLE III. sists in reacting a ketone of formula 1(p-Chlorophenyl)-1-(21-pyridyl)-3-di-

Ar,CO.CH\_.CH\_R

methylaminopropanc.

Using p-chlorophenylacetophenone in the

wherein Ar stands for phenyl or for an alkyl-, alkozy-, dialkylamino-, chloro- or bromophenyl and R stands for a dialkylamino-, piperidino-, pyrrolidino- or morpholino-group, with an organometalic 2-pyridyl compound (e.g. 2-pyridyllithium or 2-pyridyl magnesium halide) to give the carbinol

wherein Py stands for 2-pyridyl followed by 10 replacement of the hydroxyl group in the resulting carbinol by hydrogen to give the compound.

and conversion of the product, if desired, into its salts.

The step in the production of pyridyl: aliphatic amines and their salts as claimed in Claim 1 comprising the conversion of the carbinol into the corresponding halide, e.g.
 by the action of thionyl chloride and replacement of the halogen by hydrogen, e.g. by reduction with zinc dust and acetic acid, to give a compound of formula

and conversion of this product if desired, into its salts.

 The steps as claimed in either of the preceding claims in which Ar stands for phenyl or p-chlorophenyl and R for dimethylamino or N-pyrrolidyl.

4. Process for the production of saturated compounds of the formula

substantially as described with reference to each of the foregoing Examples.

5. 3-(21-Pyridyl)-3-arylpropylamines, whenever produced by the process claimed in any of the preceding claims...

6. Salts of 3-(21-pyridyI)-3 - aryl-propylamines whenever produced by the process claimed in any of Claims 1—3.

Dated this 13th day of October, 1949.

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Reference has been directed in pursuance of Section 9, sub-section (1) of the Patents Act, 1949 to Parent No. 689,234.

Leamington Spa: Printed for Her Majesty's Stationery Office, by the Courier Press.—1957. Published at The Patent Office, 25, Southampton Buildings, London, W.C.2, from which copies may be obtained.